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## Key indicators

Single-crystal X-ray study
$T=298 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.009 \AA$
$R$ factor $=0.055$
$w R$ factor $=0.135$
Data-to-parameter ratio $=13.5$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## Poly[nickel(II)- $\mu$-4,4'-bipyridine- $\mu$-terephthalato]

4,4'-Bipyridine(terephthalato)nickel(II), $\left[\mathrm{Ni}\left(\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{O}_{4}\right)\left(\mathrm{C}_{10} \mathrm{H}_{8}-\right.\right.$ $\left.\mathrm{N}_{2}\right)$, exists as poly[ $\mu_{2}$-terephthalato- $\mu_{4}$-terephthalato-bis[( $\mu_{2}-4,4^{\prime}$-bipyridine)nickel(II)]]. Two independent terephthalato groups both occupy positions of $2 / m$ symmetry; one of these groups acts as a $\mu_{4}$ bridge and coordinates four different Ni atoms, one with each of its O atoms. The other one acts as a $\mu_{2}$-bridge and serves as a bidentate chelate for two neighboring metal atoms. The Ni atom lies in a special position of $m$ site symmetry.

## Comment

4,4'-Bipyridine(terephthalato)nickel(II), $\left[\mathrm{Ni}\left(\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{O}_{4}\right)\left(\mathrm{C}_{10} \mathrm{H}_{8^{-}}\right.\right.$ $\left.\mathrm{N}_{2}\right)$ ], exists as poly $\left[\mu_{2}\right.$-terephthalato- $\mu_{4}$-terephthalato-bis[( $\mu_{2}-4,4^{\prime}$-bipyridine)nickel(II)]]. Two independent terephthalato groups both occupy positions of $2 / \mathrm{m}$ symmetry; one of these groups acts as a $\mu_{4}$ bridge and coordinates four different Ni atoms, one with each of its O atoms. The other one acts as a $\mu_{2}$-bridge and serves as a bidentate chelate for two neighboring metal atoms (Fig. 1). Consequently, the terephthalatenickel framework forms infinite two-dimensional layers parallel to the $a b$ plane of the crystal. The layers are linked into a three-dimensional network through the $\mu_{2}$-bridging 4,4'-bipyridine ligands occupying a special position across the mirror plane. The compound is isostructural with the published cobalt analog, whose detailed description (Tao et al., 2000) also applies to the title compound.

(I)

## Experimental

The compound was synthesized hydrothermally from nickel nitrate hexahydrate ( $0.29 \mathrm{~g}, 1 \mathrm{mmol}$ ), terephthalic acid ( $0.17 \mathrm{~g}, 1 \mathrm{mmol}$ ), 4,4'bipyridine dihydrochloride ( $0.23 \mathrm{~g}, 1 \mathrm{mmol}$ ) and sodium hydroxide $(0.16 \mathrm{~g}, 4 \mathrm{mmol})$ in water $(18 \mathrm{ml})$. The mixture was placed in a 20 ml Teflon-lined stainless-steel vessel, which was heated at 493 K for 100 h . The vessel was cooled to room temperature at a rate of 6 K $\mathrm{h}^{-1}$. The product was isolated in $25 \%$ yield.

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## Crystal data

$\left[\mathrm{Ni}\left(\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{O}_{4}\right)\left(\mathrm{C}_{10} \mathrm{H}_{8} \mathrm{~N}_{2}\right)\right] \quad D_{x}=1.512 \mathrm{Mg} \mathrm{m}^{-3}$
$M_{r}=379.01$
Monoclinic, $C 2 / m$
$a=16.610$ (3) Å
$b=10.222$ (2) $\AA$
$c=11.237$ (2) $\AA$
$\beta=119.22(3)^{\circ}$
$V=1665.1(5) \AA^{3}$
$Z=4$

## Data collection

| Enraf-Nonius CAD-4 | $R_{\text {int }}=0.038$ |
| :--- | :--- |
| $\quad$ diffractometer | $\theta_{\max }=26.0^{\circ}$ |
| $\omega$ scans | $h=0 \rightarrow 20$ |
| Absorption correction: $\psi$ scan | $k=0 \rightarrow 12$ |
| $\quad$ (North et al., 1968) | $l=-13 \rightarrow 12$ |
| $T_{\text {min }}=0.696, T_{\text {max }}=0.877$ | 3 standard reflections |
| 1790 measured reflections | frequency: 60 min |
| 1728 independent reflections | intensity decay: none |

1312 reflections with $I>2 \sigma(I)$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.055$
$w R\left(F^{2}\right)=0.135$
$S=1.03$
1728 reflections
128 parameters
H-atom parameters constrained

$$
\begin{aligned}
& w=1 / {\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0649 P)^{2}\right.} \\
&+1.5719 P] \\
& \text { where } P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.67 \mathrm{e}^{-3} \AA^{-3} \\
& \Delta \rho_{\min }=-0.73 \mathrm{e} \AA^{-3}
\end{aligned}
$$

Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right)$.

| Ni1-O1 | $1.999(3)$ | Ni1-N1 | $2.076(5)$ |
| :--- | ---: | :--- | ---: |
| Ni1-O2 | $2.149(3)$ |  |  |
| $\mathrm{O} 1-\mathrm{Ni} 1-\mathrm{O} 1^{\mathrm{i}}$ | $111.4(2)$ | $\mathrm{O} 2-\mathrm{Ni} 1-\mathrm{O} 2^{\mathrm{i}}$ | $61.3(2)$ |
| $\mathrm{O} 1-\mathrm{Ni} 1-\mathrm{O} 2$ | $154.7(1)$ | $\mathrm{O} 2-\mathrm{Ni} 1-\mathrm{N} 1$ | $89.5(2)$ |
| $\mathrm{O} 1-\mathrm{Ni} 1-\mathrm{O} 2^{\mathrm{i}}$ | $93.5(1)$ | $\mathrm{O} 2-\mathrm{Ni} 1-\mathrm{N} 2^{\mathrm{ii}}$ | $87.3(2)$ |
| $\mathrm{O} 1-\mathrm{Ni} 1-\mathrm{N} 1$ | $93.4(1)$ | $\mathrm{N} 1-\mathrm{Ni} 1-\mathrm{N} 2^{i i}$ | $176.3(2)$ |
| $\mathrm{O} 1-\mathrm{Ni} 1-\mathrm{N} 2^{\mathrm{ii}}$ | $88.6(1)$ |  |  |

Symmetry codes: (i) $x, 1-y, z$; (ii) $x, y, 1+z$.
The aromatic portion of one of the terephthalate groups (at the Wyckoff $2 d$ site of $2 / m$ symmetry), made up of atoms C5, C6 and C7, showed large displacement parameters for atoms C 6 and C 7 . H atoms were positioned geometrically $(\mathrm{C}-\mathrm{H}=0.93 \AA)$ and included in the subsequent refinement in the riding motion approximation with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$.

Data collection: CAD-4 Software (Enraf-Nonius, 1988); cell refinement: CAD-4 Software; data reduction: XCAD4 (Harms, 1997);


Figure 1
ORTEPII (Johnson, 1976) plot of a fragment of the structure of (I), with ellipsoids drawn at the $50 \%$ probability level. H atoms are drawn as spheres of arbitrary radii. [Symmetry codes: (i) $x, 1-y, z$; (ii) $x, y, 1+z$.]
program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: $S H E L X L 97$.

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