# metal-organic papers

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#### Key indicators

Single-crystal X-ray study T = 298 K Mean  $\sigma$ (C–C) = 0.009 Å R factor = 0.055 wR factor = 0.135 Data-to-parameter ratio = 13.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

## **Poly**[nickel(II)-μ-4,4'-bipyridine-μ-terephthalato]

4,4'-Bipyridine(terephthalato)nickel(II), [Ni( $C_8H_4O_4$ )( $C_{10}H_8$ - $N_2$ )], exists as poly[ $\mu_2$ -terephthalato- $\mu_4$ -terephthalato-bis-[( $\mu_2$ -4,4'-bipyridine)nickel(II)]]. Two independent terephthalato groups both occupy positions of 2/m symmetry; one of these groups acts as a  $\mu_4$  bridge and coordinates four different Ni atoms, one with each of its O atoms. The other one acts as a  $\mu_2$ -bridge and serves as a bidentate chelate for two neighboring metal atoms. The Ni atom lies in a special position of m site symmetry.

#### Comment

4,4'-Bipyridine(terephthalato)nickel(II), [Ni(C<sub>8</sub>H<sub>4</sub>O<sub>4</sub>)(C<sub>10</sub>H<sub>8</sub>-N<sub>2</sub>)], exists as poly[ $\mu_2$ -terephthalato- $\mu_4$ -terephthalato-bis-[( $\mu_2$ -4,4'-bipyridine)nickel(II)]]. Two independent terephthalato groups both occupy positions of 2/m symmetry; one of these groups acts as a  $\mu_4$  bridge and coordinates four different Ni atoms, one with each of its O atoms. The other one acts as a  $\mu_2$ -bridge and serves as a bidentate chelate for two neighboring metal atoms (Fig. 1). Consequently, the terephthalatenickel framework forms infinite two-dimensional layers parallel to the *ab* plane of the crystal. The layers are linked into a three-dimensional network through the  $\mu_2$ -bridging 4,4'-bipyridine ligands occupying a special position across the mirror plane. The compound is isostructural with the published cobalt analog, whose detailed description (Tao *et al.*, 2000) also applies to the title compound.



#### **Experimental**

The compound was synthesized hydrothermally from nickel nitrate hexahydrate (0.29 g, 1 mmol), terephthalic acid (0.17 g, 1 mmol), 4,4'bipyridine dihydrochloride (0.23 g, 1 mmol) and sodium hydroxide (0.16 g, 4 mmol) in water (18 ml). The mixture was placed in a 20 ml Teflon-lined stainless-steel vessel, which was heated at 493 K for 100 h. The vessel was cooled to room temperature at a rate of 6 K  $h^{-1}$ . The product was isolated in 25% yield.

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## Crystal data

 $\begin{bmatrix} \text{Ni}(\text{C}_8\text{H}_4\text{O}_4)(\text{C}_{10}\text{H}_8\text{N}_2) \end{bmatrix} \\ M_r = 379.01 \\ \text{Monoclinic, } C2/m \\ a = 16.610 \text{ (3)} \text{ Å} \\ b = 10.222 \text{ (2)} \text{ Å} \\ c = 11.237 \text{ (2)} \text{ Å} \\ \beta = 119.22 \text{ (3)}^{\circ} \\ V = 1665.1 \text{ (5)} \text{ Å}^3 \\ Z = 4 \\ \end{bmatrix}$ 

### Data collection

Enraf–Nonius CAD-4 diffractometer  $\omega$  scans Absorption correction:  $\psi$  scan (North *et al.*, 1968)  $T_{min} = 0.696$ ,  $T_{max} = 0.877$ 1790 measured reflections 1728 independent reflections 1312 reflections with  $I > 2\sigma(I)$ 

## Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.055$   $wR(F^2) = 0.135$  S = 1.031728 reflections 128 parameters H-atom parameters constrained

#### Table 1

Selected geometric parameters (Å, °).

Ni1-O1 Ni1-O2	1.999 (3) 2.149 (3)	Ni1-N1	2.076 (5)
$\begin{array}{c} 01 - \text{Ni}1 - 01^{\text{i}} \\ 01 - \text{Ni}1 - 02 \\ 01 - \text{Ni}1 - 02^{\text{i}} \\ 01 - \text{Ni}1 - \text{N1} \\ 01 - \text{Ni}1 - \text{N2}^{\text{ii}} \end{array}$	111.4 (2) 154.7 (1) 93.5 (1) 93.4 (1) 88.6 (1)	$O2-Ni1-O2^{i}$ O2-Ni1-N1 $O2-Ni1-N2^{ii}$ $N1-Ni1-N2^{ii}$	61.3 (2) 89.5 (2) 87.3 (2) 176.3 (2)

 $D_x = 1.512 \text{ Mg m}^{-3}$ 

Cell parameters from 25

 $0.18 \times 0.18 \times 0.11 \text{ mm}$ 

Mo  $K\alpha$  radiation

reflections

 $\mu = 1.19 \text{ mm}^{-1}$ 

T = 298 (2) K

Block, green

 $\begin{aligned} R_{\rm int} &= 0.038\\ \theta_{\rm max} &= 26.0^\circ\\ h &= 0 \rightarrow 20 \end{aligned}$ 

 $k=0\rightarrow 12$ 

 $l = -13 \rightarrow 12$ 

3 standard reflections

frequency: 60 min

intensity decay: none

 $w = 1/[\sigma^2(F_o^2) + (0.0649P)^2]$ 

where  $P = (F_o^2 + 2F_c^2)/3$ 

+ 1.5719P]

 $\begin{array}{l} (\Delta/\sigma)_{\rm max} = 0.001 \\ \Delta\rho_{\rm max} = 0.67 \ {\rm e} \ {\rm \AA}^{-3} \end{array}$ 

 $\Delta \rho_{\rm min} = -0.73 \text{ e} \text{ \AA}^{-3}$ 

 $\theta = 14.0 - 16.0^{\circ}$ 

Symmetry codes: (i) x, 1 - y, z; (ii) x, y, 1 + z.

The aromatic portion of one of the terephthalate groups (at the Wyckoff 2*d* site of 2/*m* symmetry), made up of atoms C5, C6 and C7, showed large displacement parameters for atoms C6 and C7. H atoms were positioned geometrically (C–H = 0.93 Å) and included in the subsequent refinement in the riding motion approximation with  $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$ .

Data collection: *CAD-4 Software* (Enraf–Nonius, 1988); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms, 1997);



#### Figure 1

*ORTEPII* (Johnson, 1976) plot of a fragment of the structure of (I), with ellipsoids drawn at the 50% probability level. H atoms are drawn as spheres of arbitrary radii. [Symmetry codes: (i) x, 1 - y, z; (ii) x, y, 1 + z.]

program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ORTEP*II (Johnson, 1976); software used to prepare material for publication: *SHELXL*97.

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